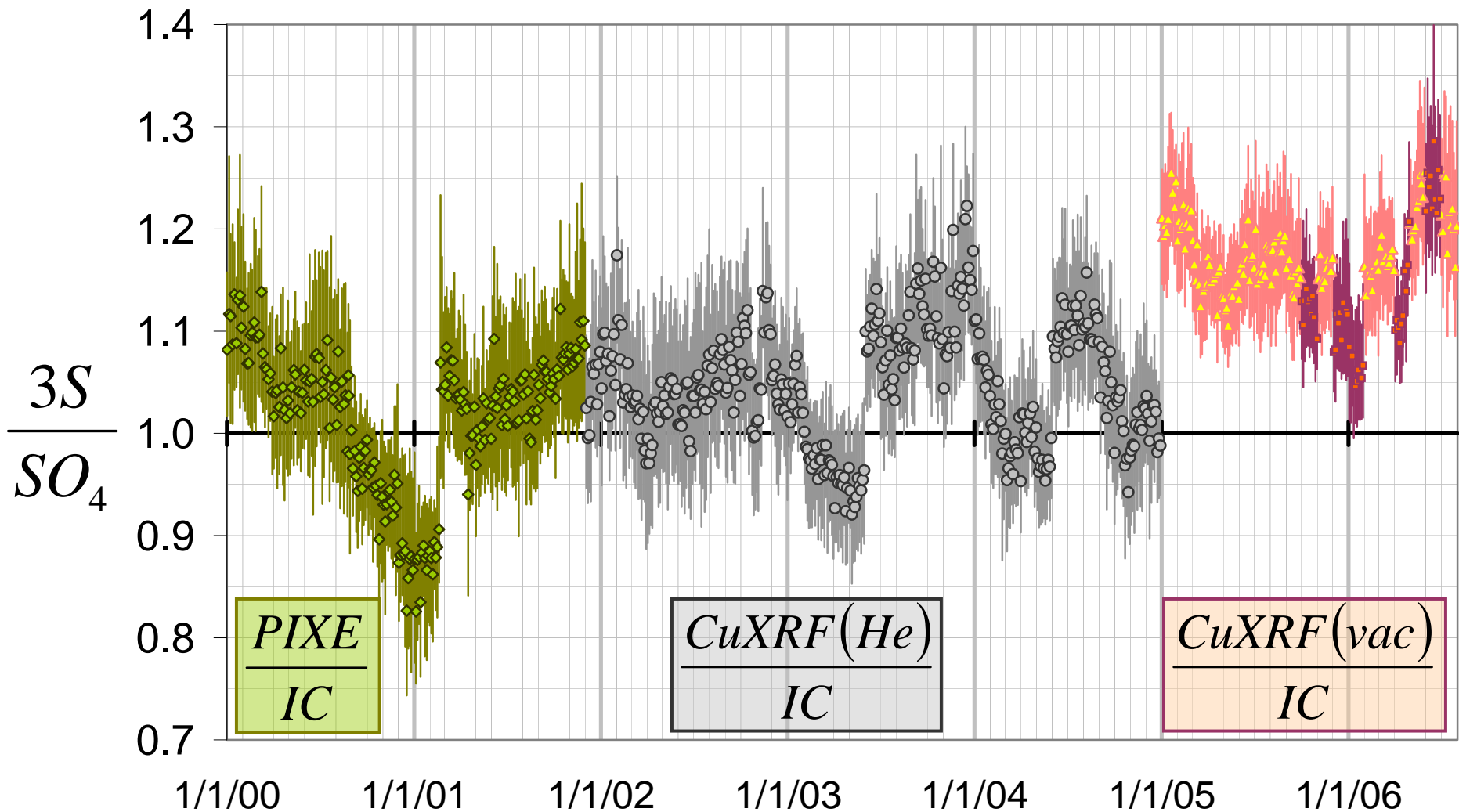


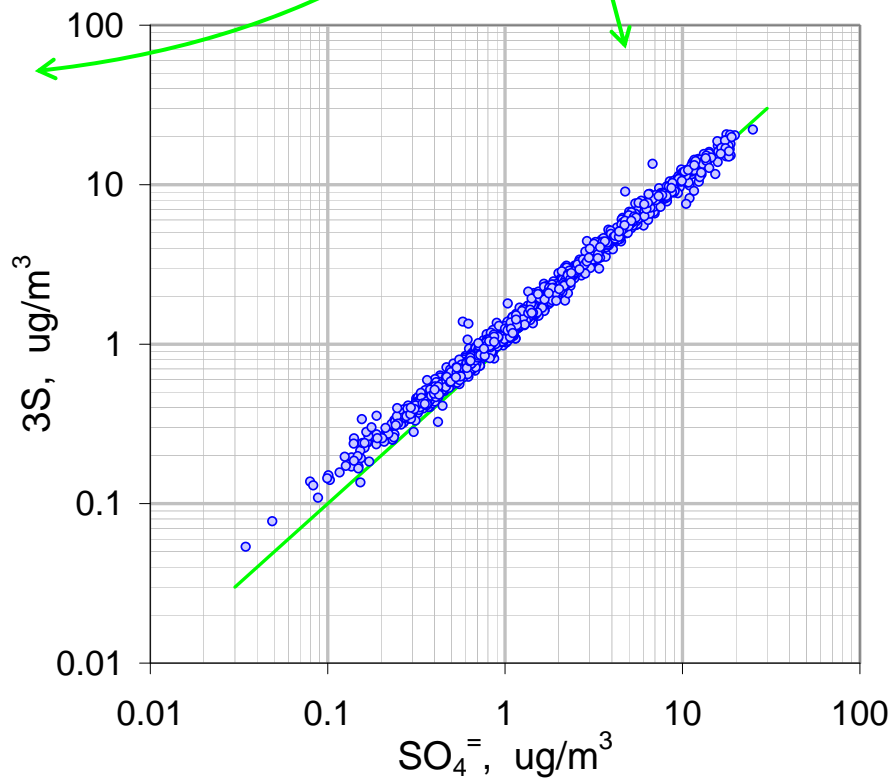
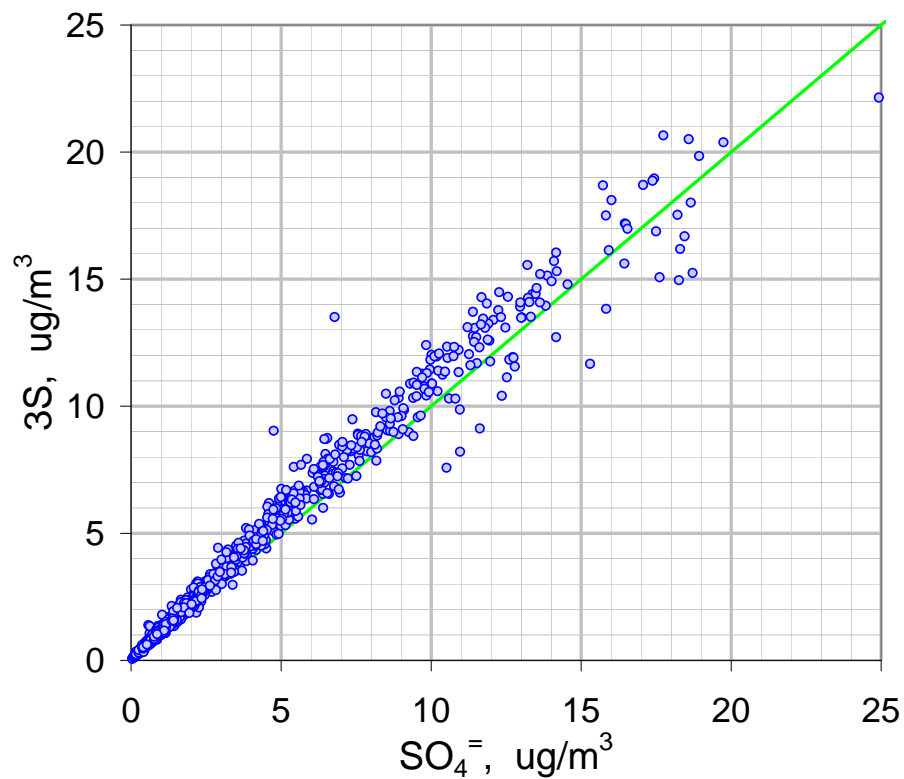
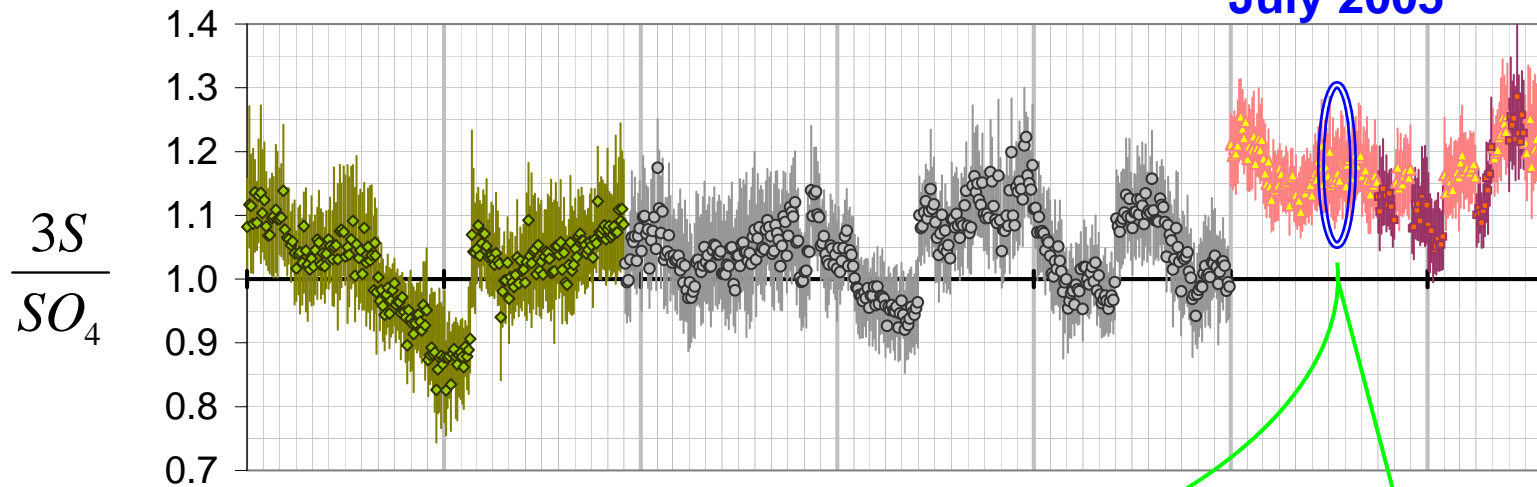
A year's progress in XRF

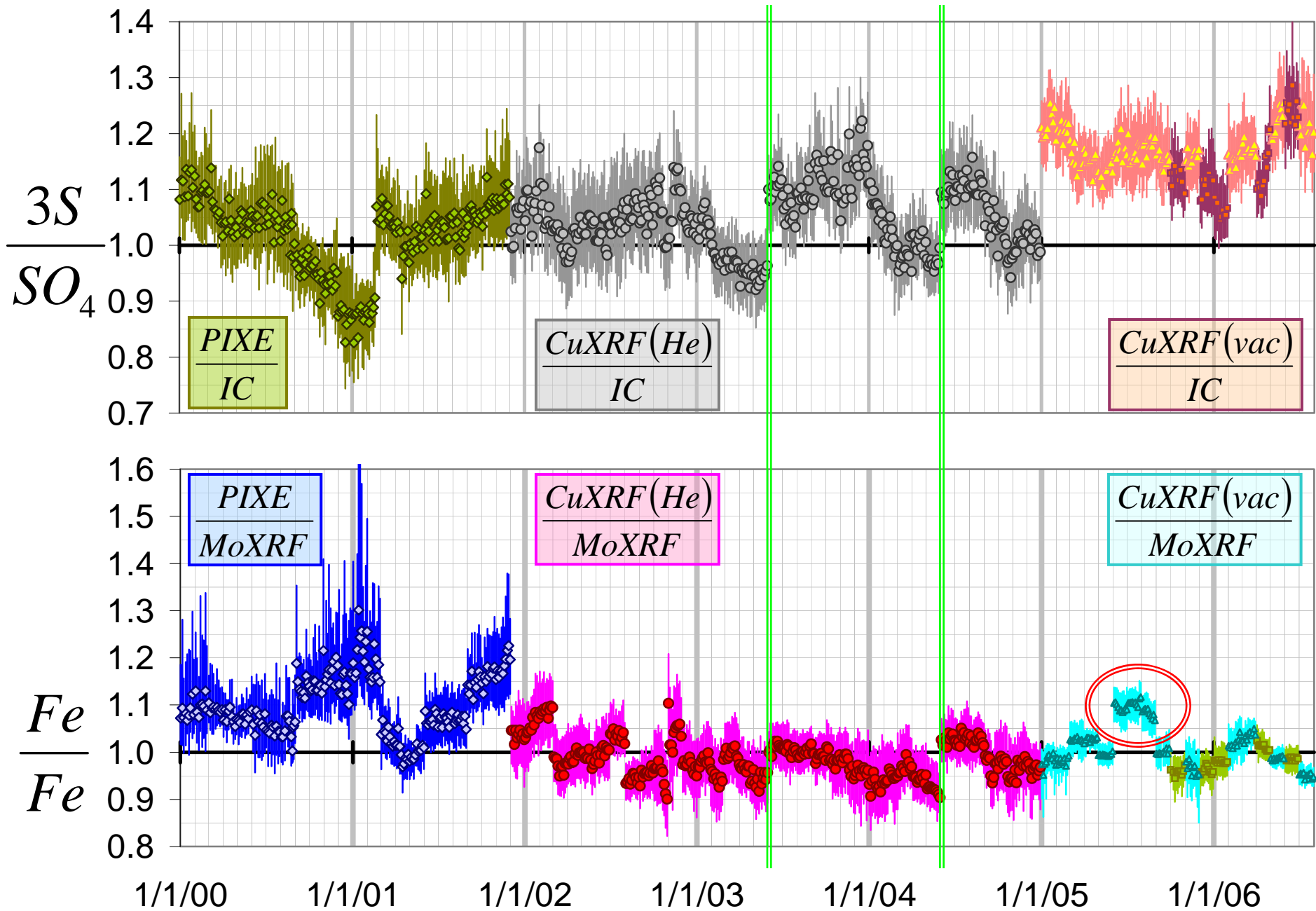
- Performance
- Documentation
- Understanding

MEDIAN AND INTERQUARTILE RANGE ENTIRE NETWORK, ALL DETECTED CONCENTRATIONS



July 2005





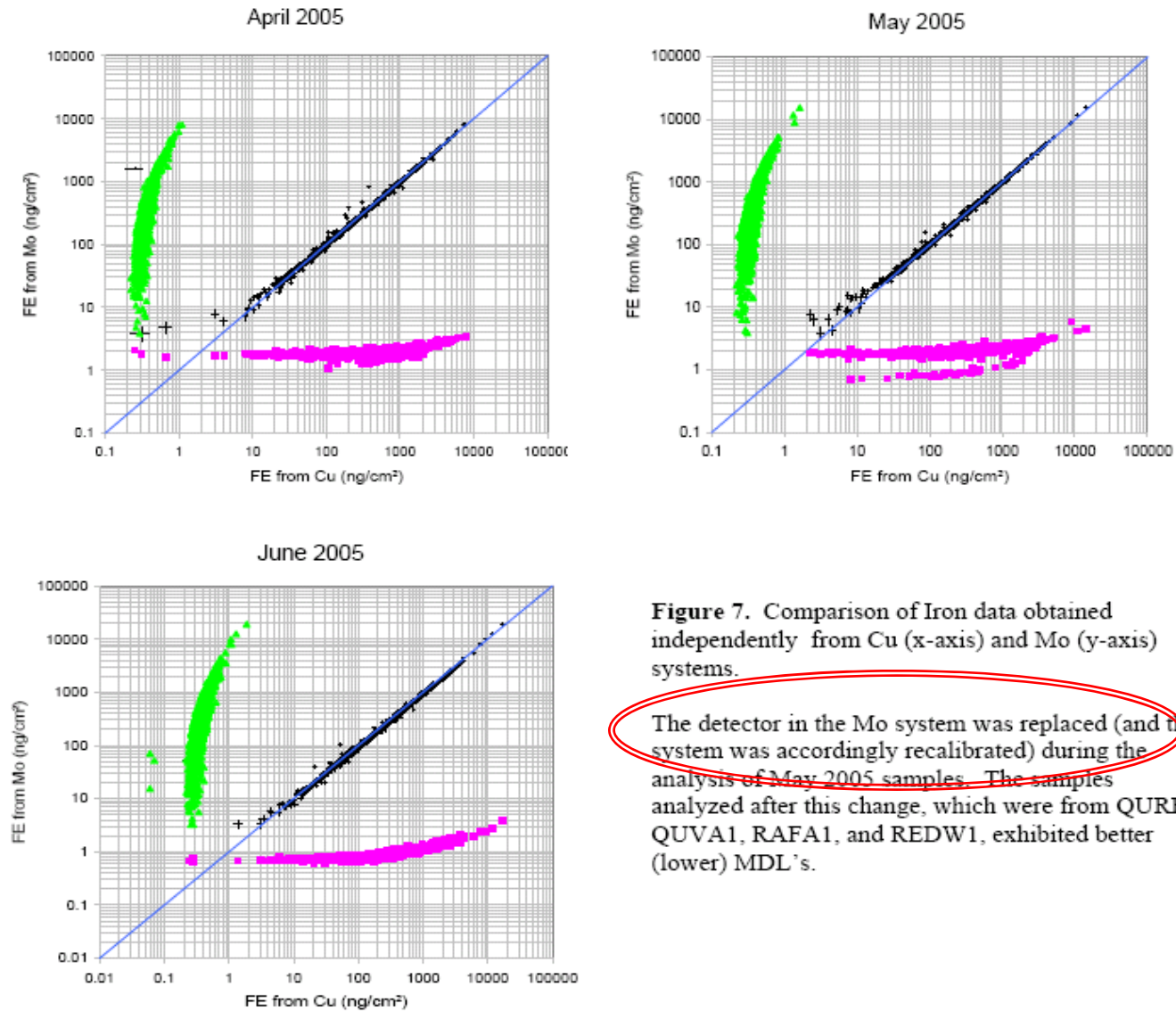
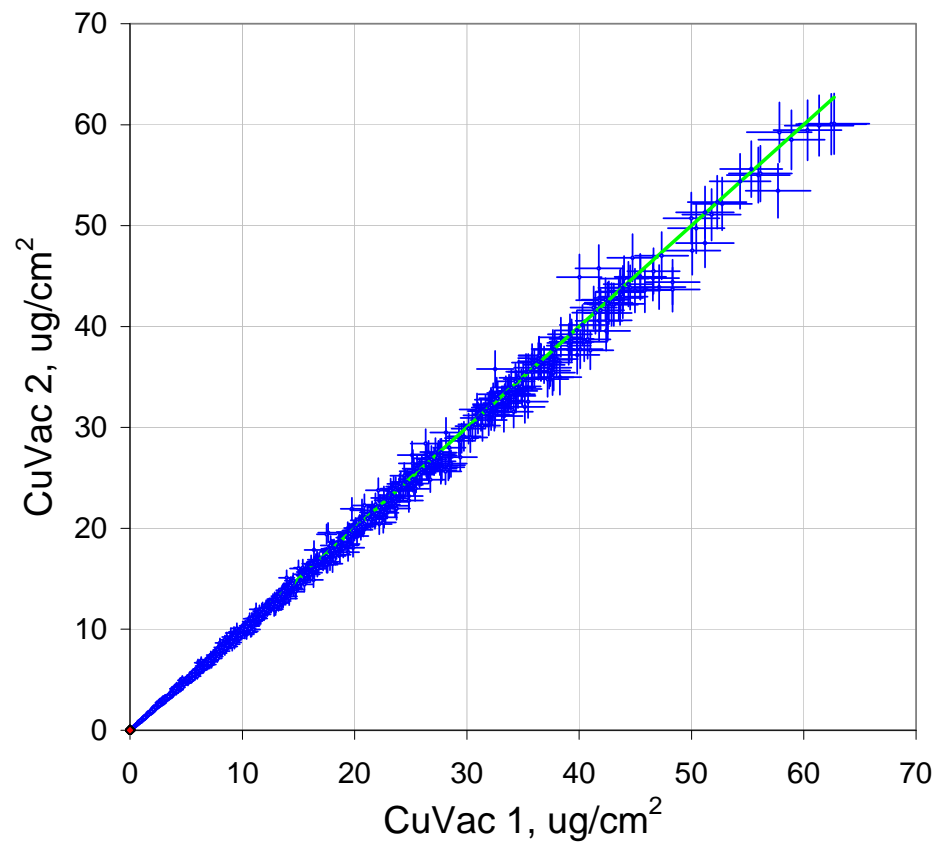
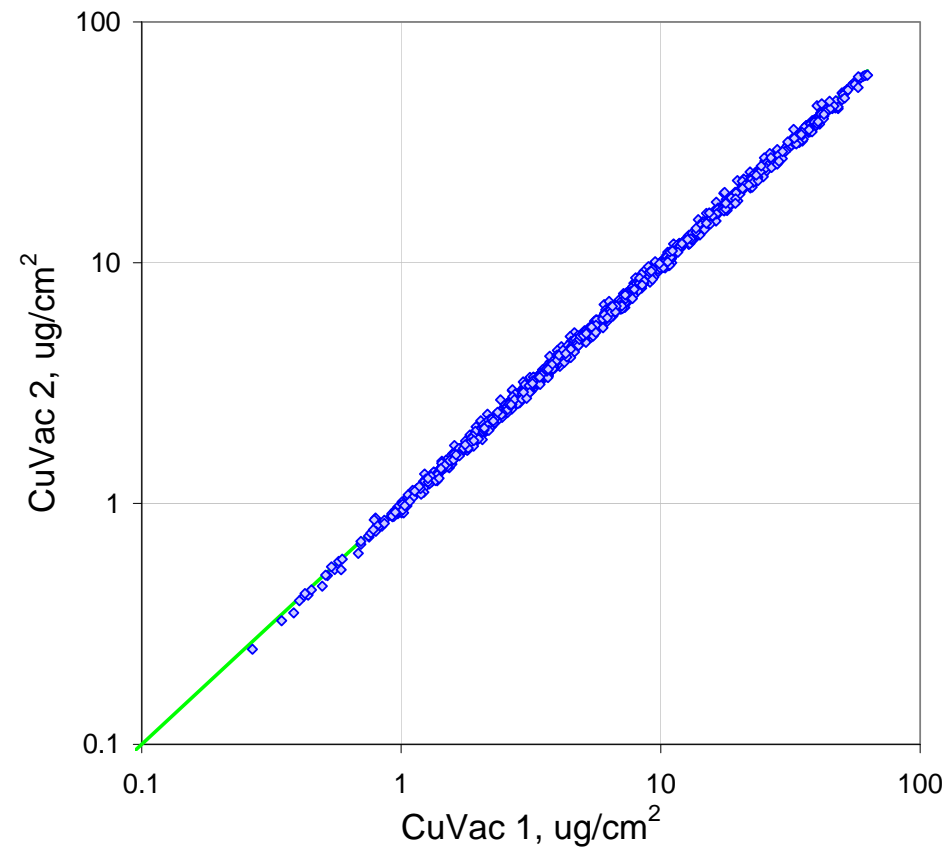


Figure 7. Comparison of Iron data obtained independently from Cu (x-axis) and Mo (y-axis) systems.

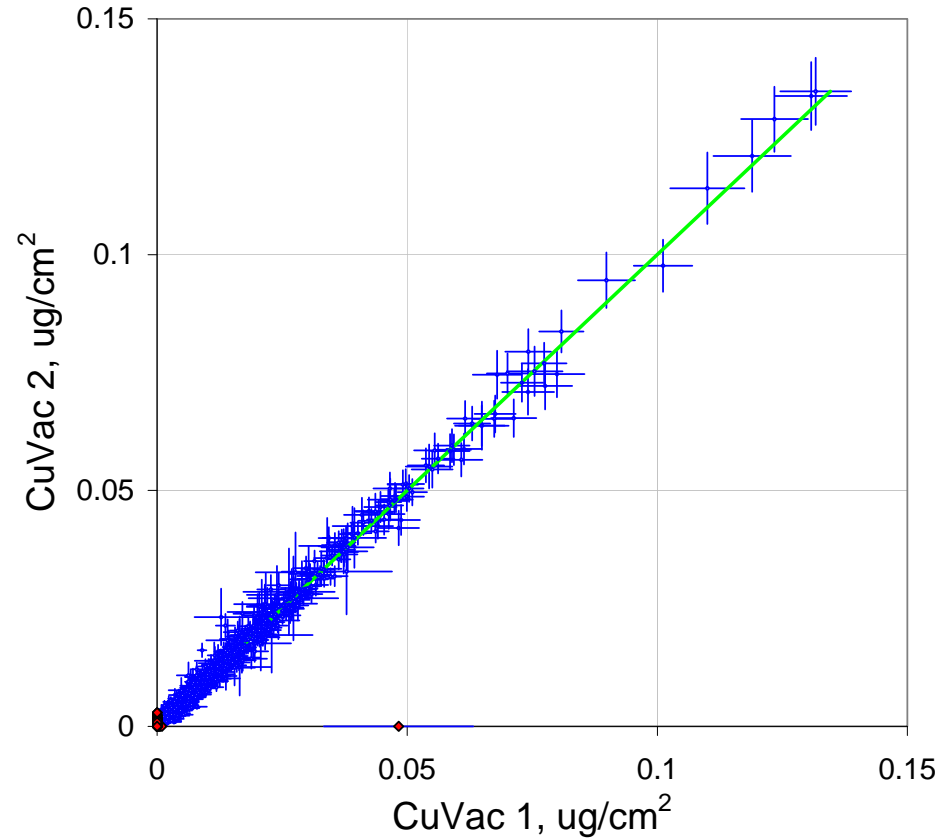
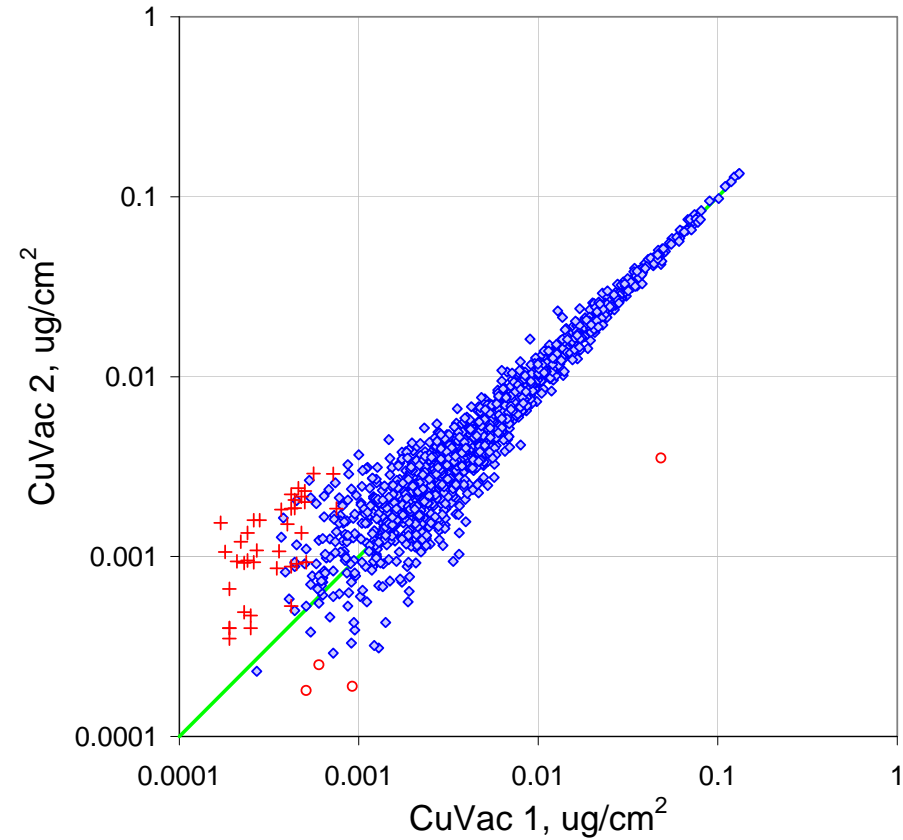
The detector in the Mo system was replaced (and the system was accordingly recalibrated) during the analysis of May 2005 samples. The samples analyzed after this change, which were from QURE1, QUVA1, RAFA1, and REDW1, exhibited better (lower) MDL's.

S, July 2005



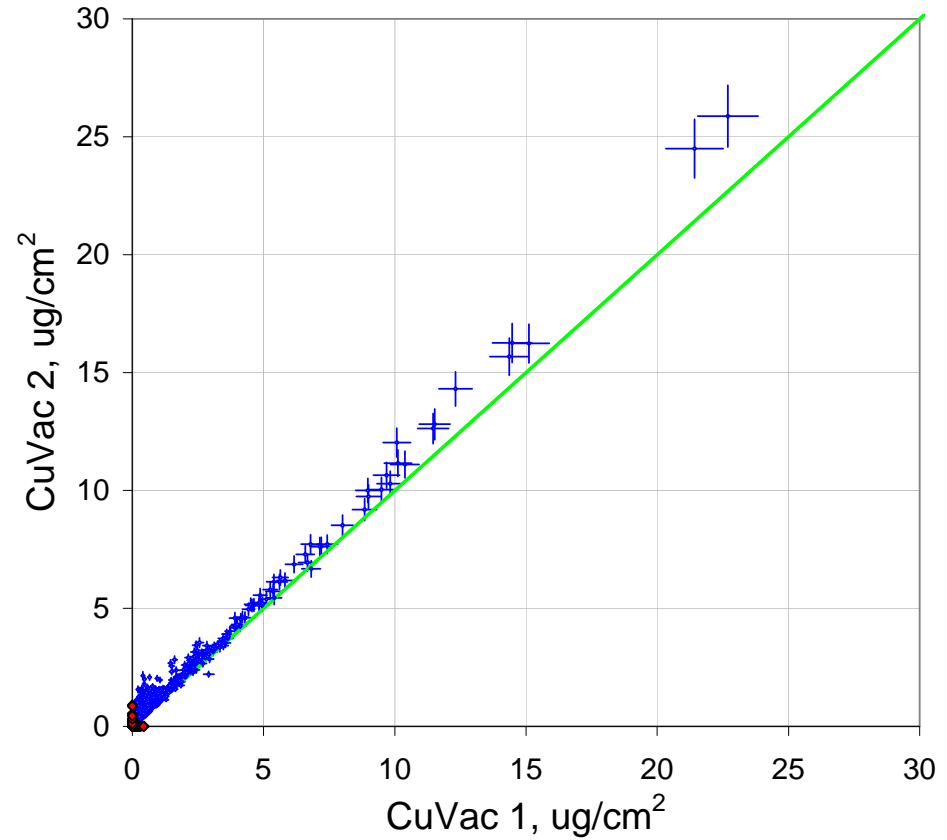
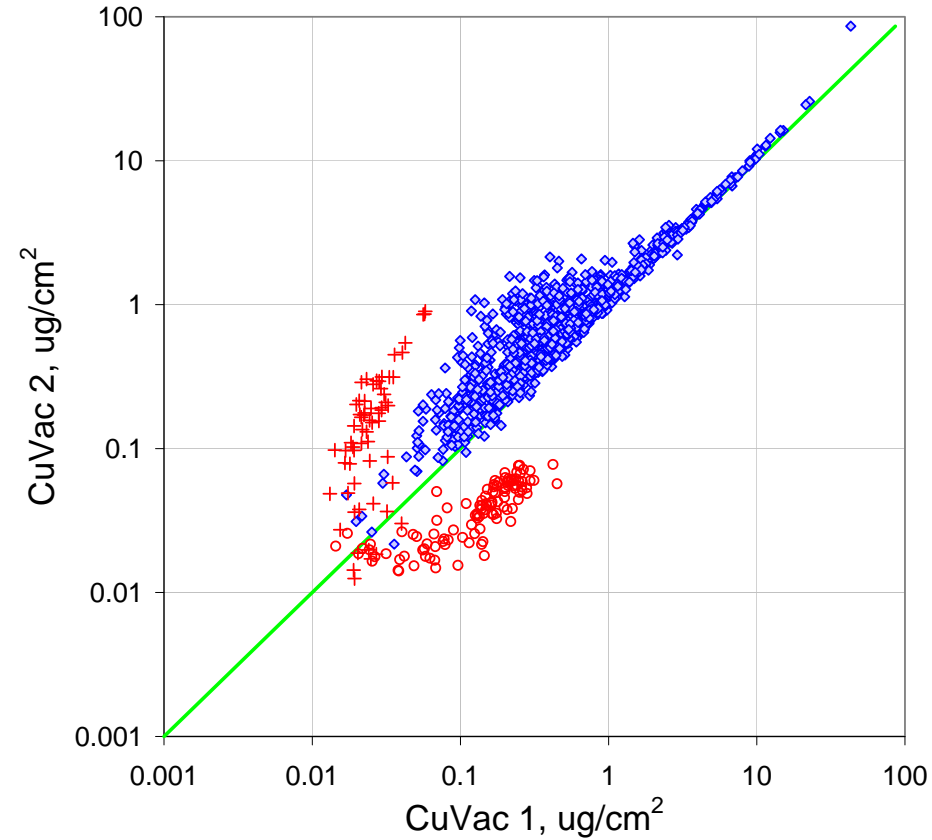
The introduction of a duplicate Cu-anode system made new comparisons possible.

V, July 2005



Vanadium is closer than sulfur to the detection limit, but the uncertainty is similar in both systems and is adequately characterized.

Al, July 2005



The “interference-free” detection limit reported for Al is known to be unrepresentative of actual performance,

as is now documented:

http://vista.cira.colostate.edu/improve/Data/QA_QC/Advisory/da0010/da0010_Almdl.pdf

Posting type	Advisory
Subject	Elemental concentrations above the MDL can go undetected
Module/Species	A/ Al
Sites	entire network
Period	entire record
Recommendation	Estimate undetected Al from Si (e.g. Eldred, 2003)

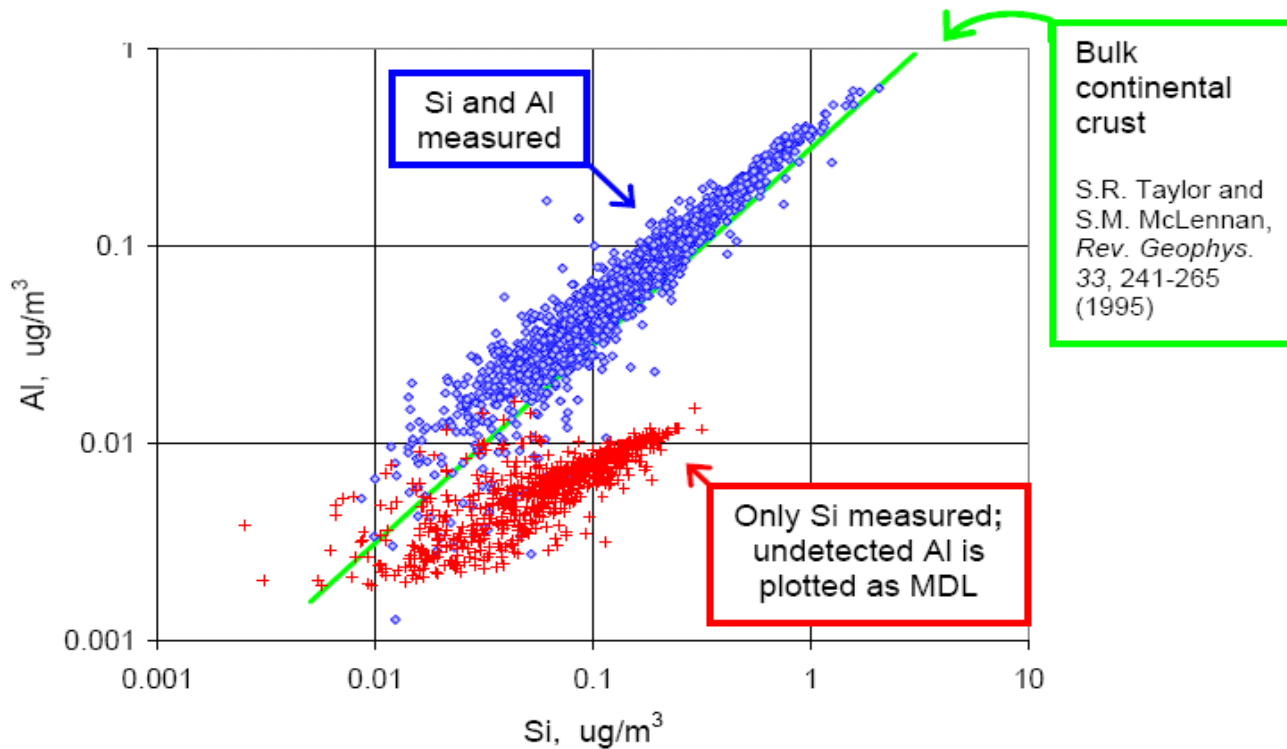
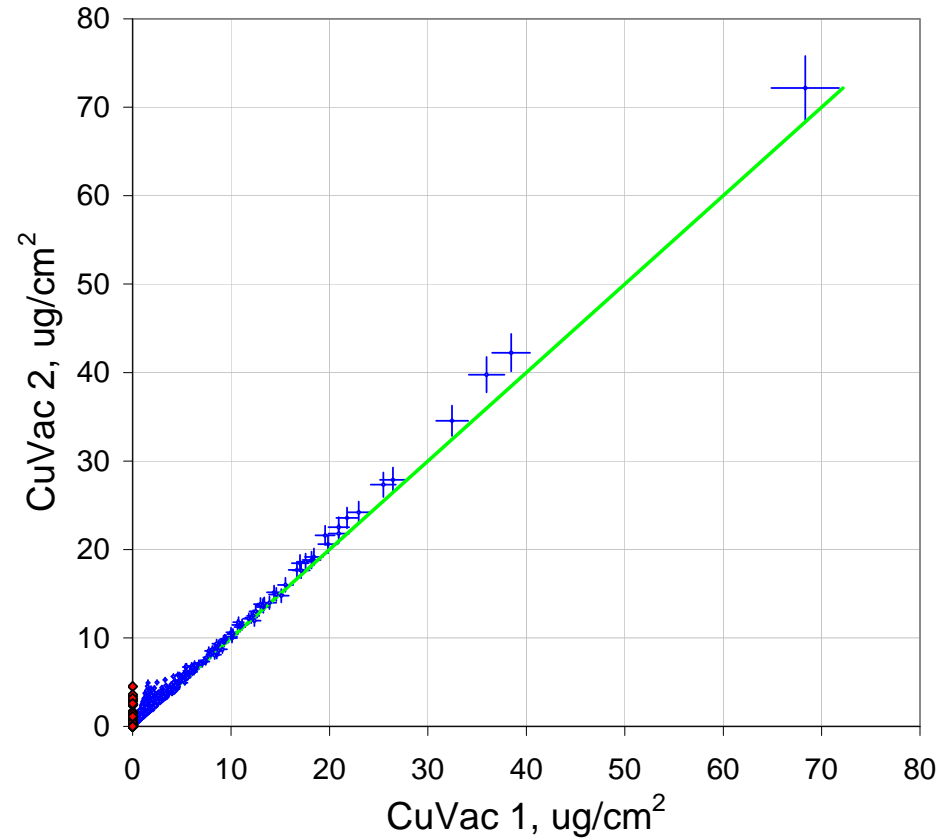
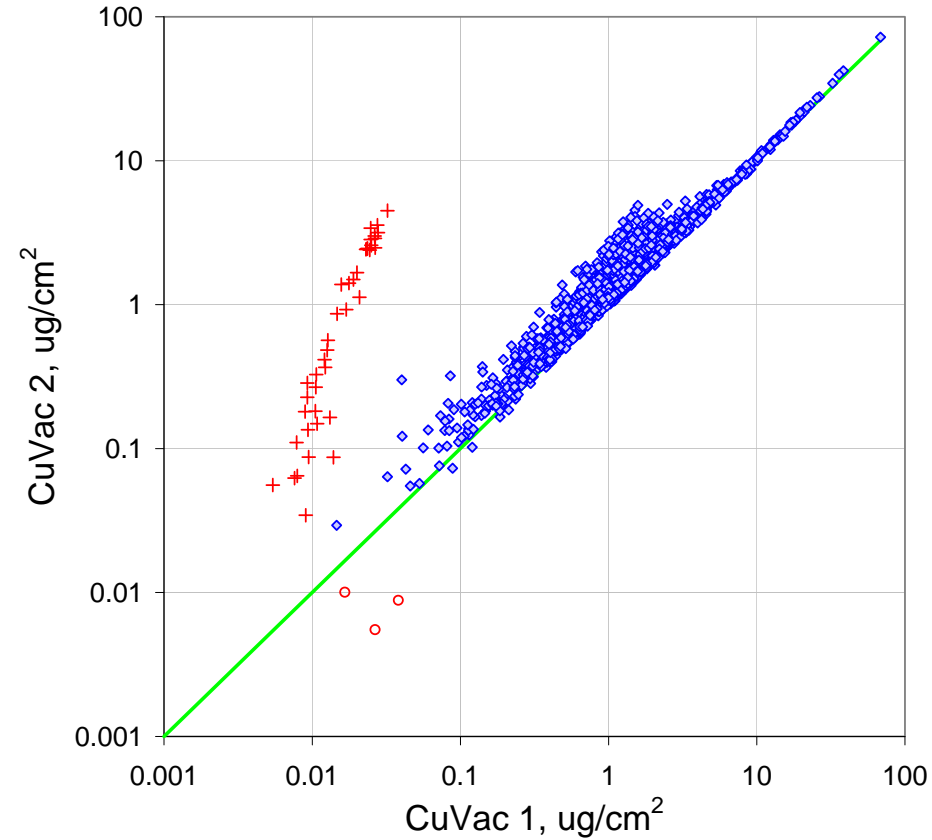
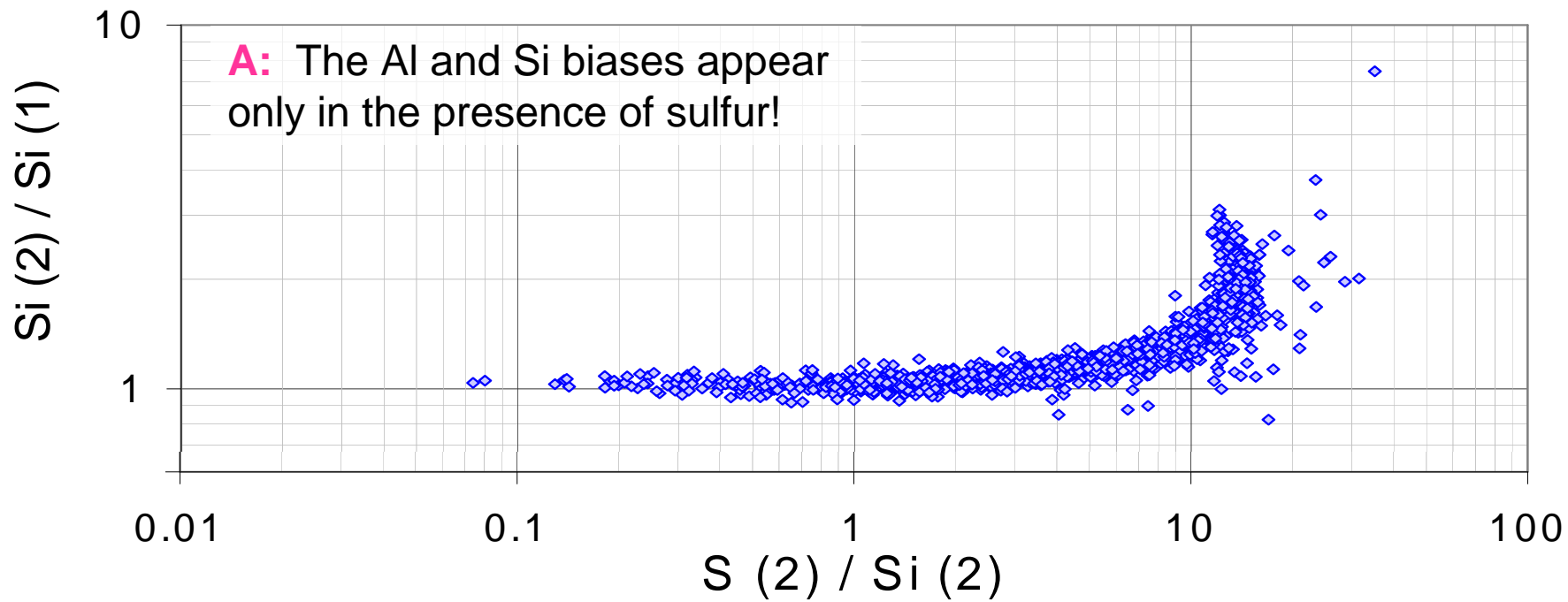
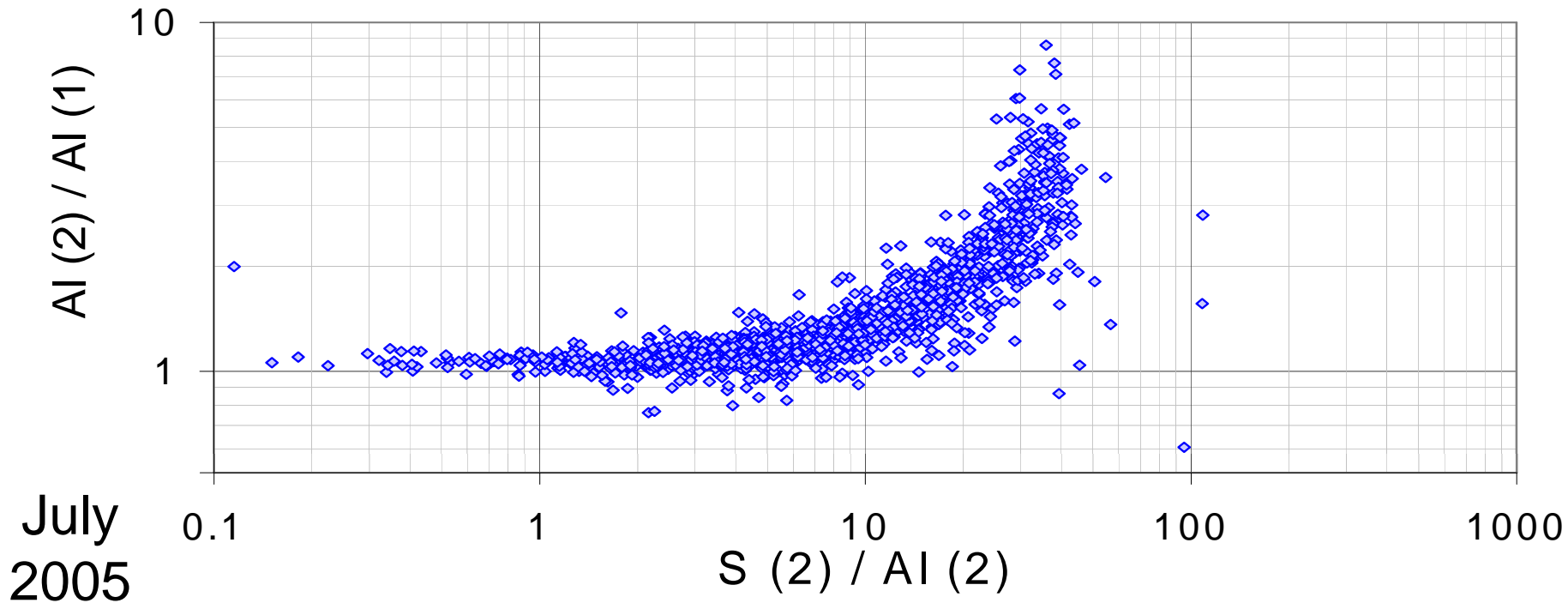


Figure 1. Aluminum and silicon concentrations from all non-urban sites in the network during March-April 2004.

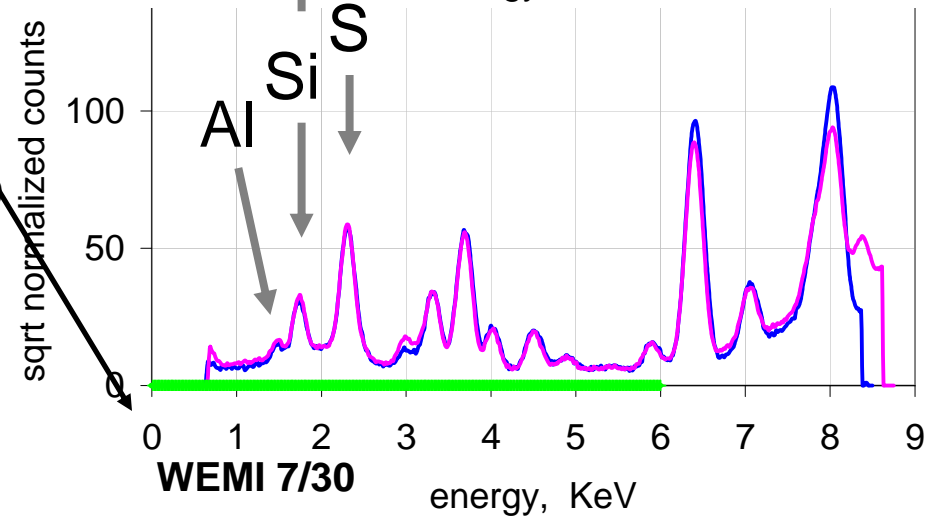
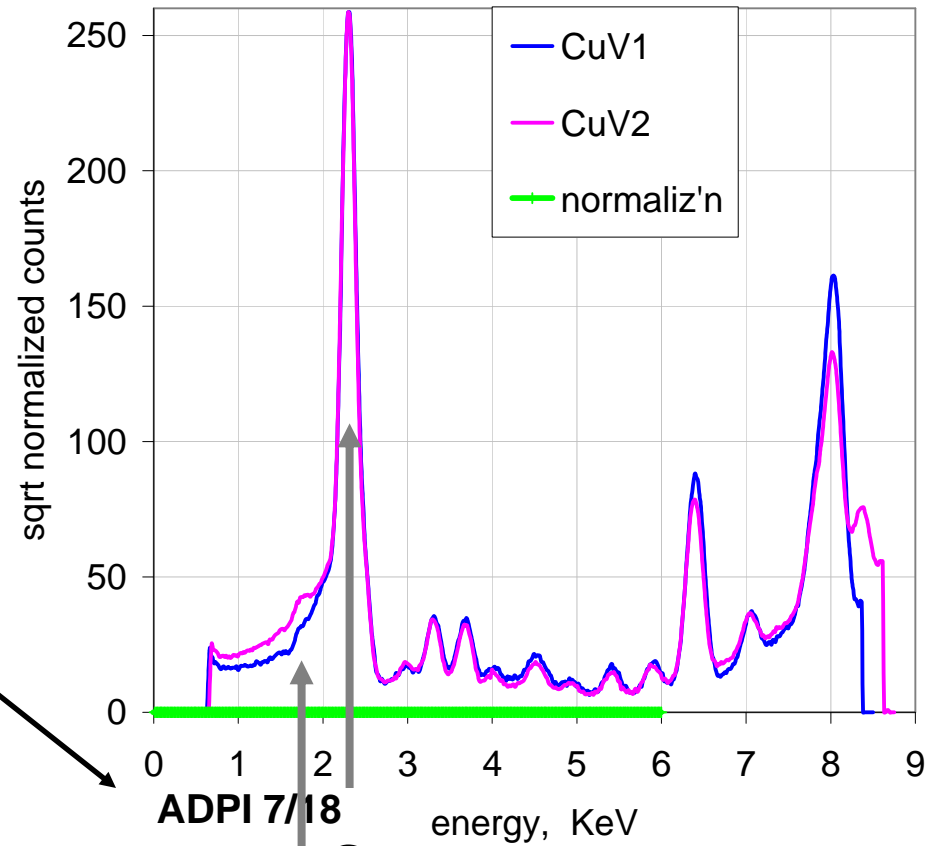
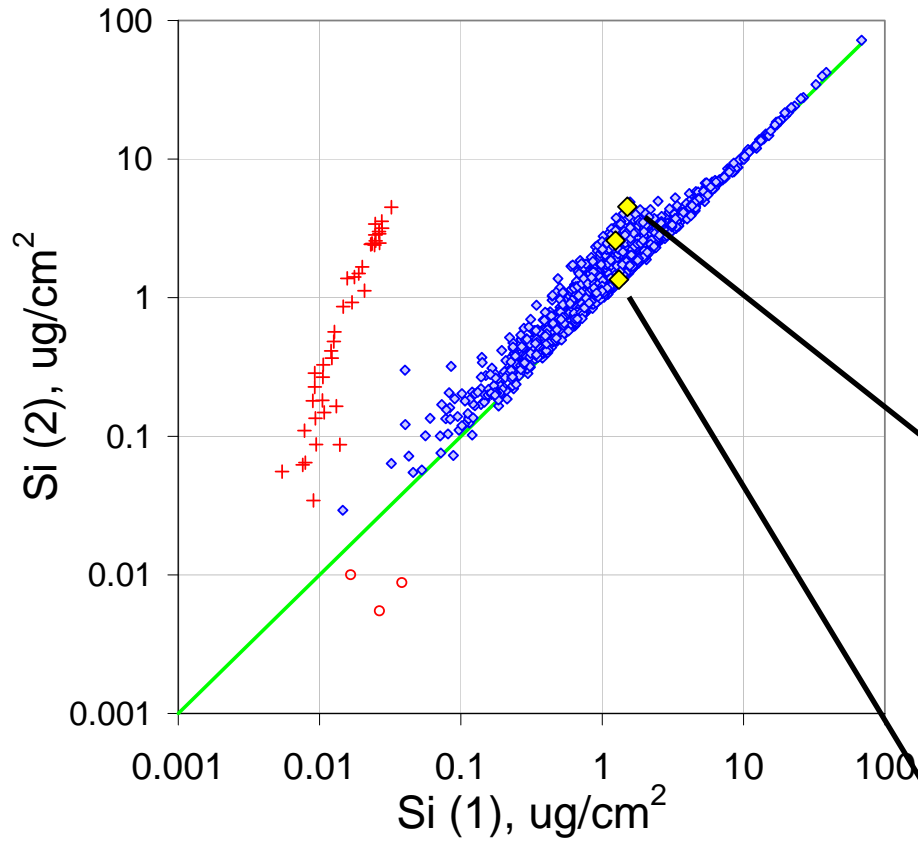
Si, July 2005



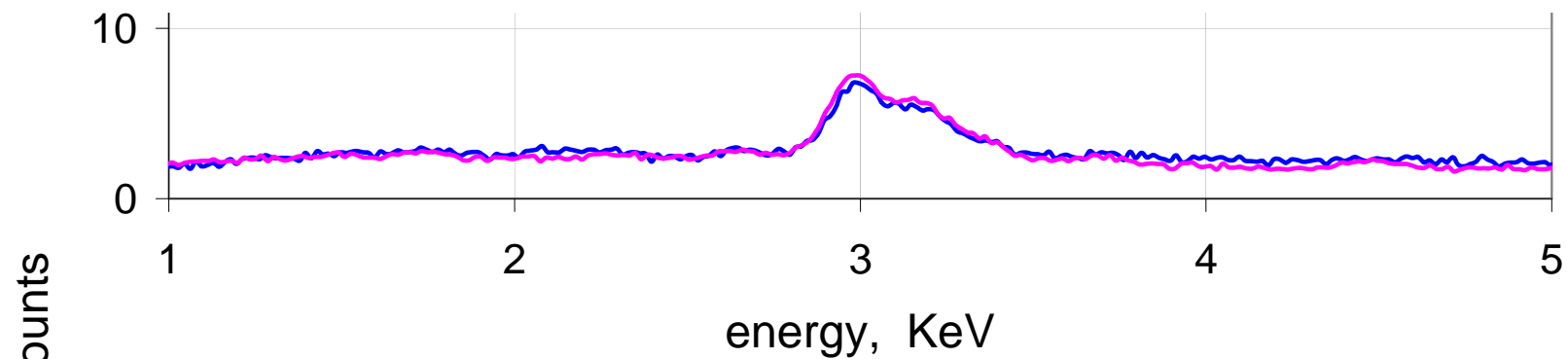
Q: How can the two systems show such relative bias for Si and Al when they are both calibrated to the same standards?



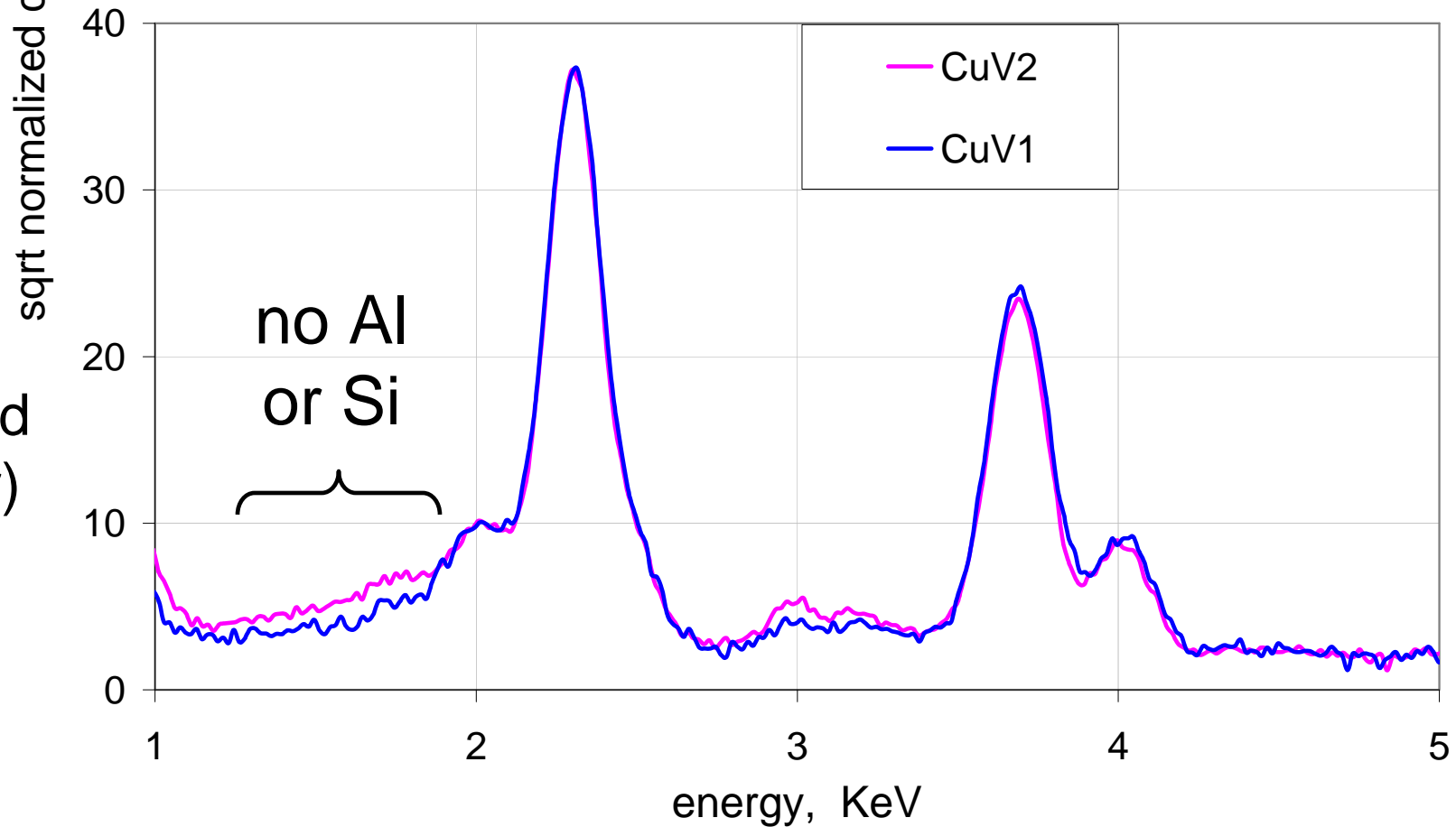
July 2005



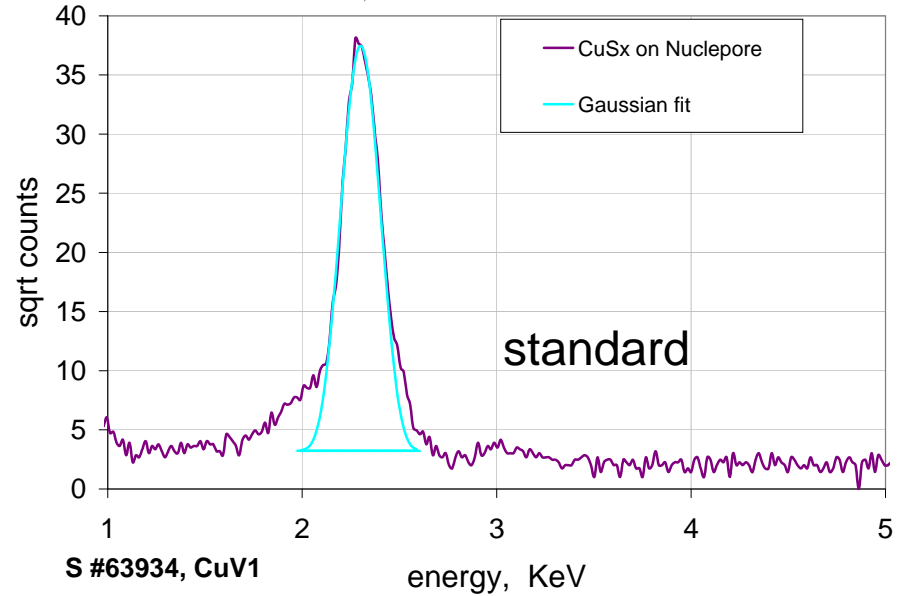
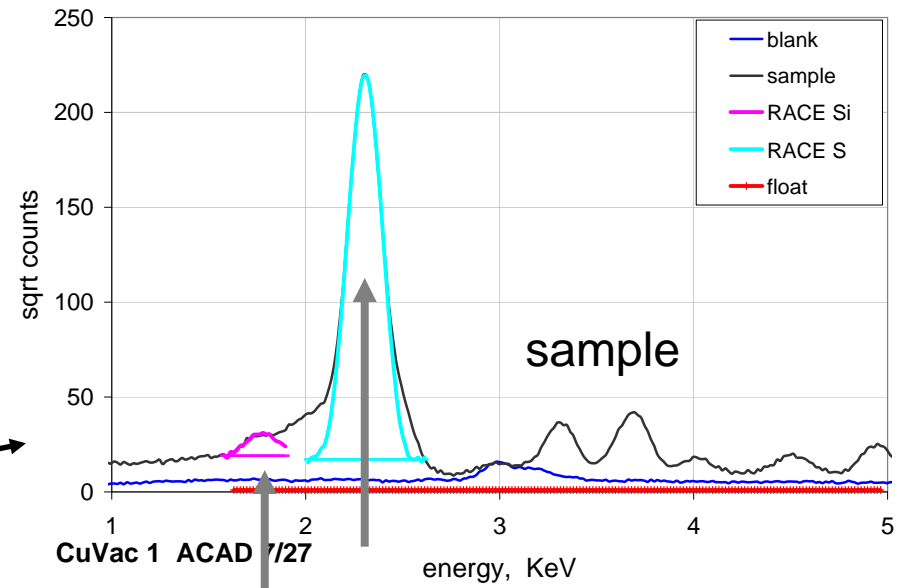
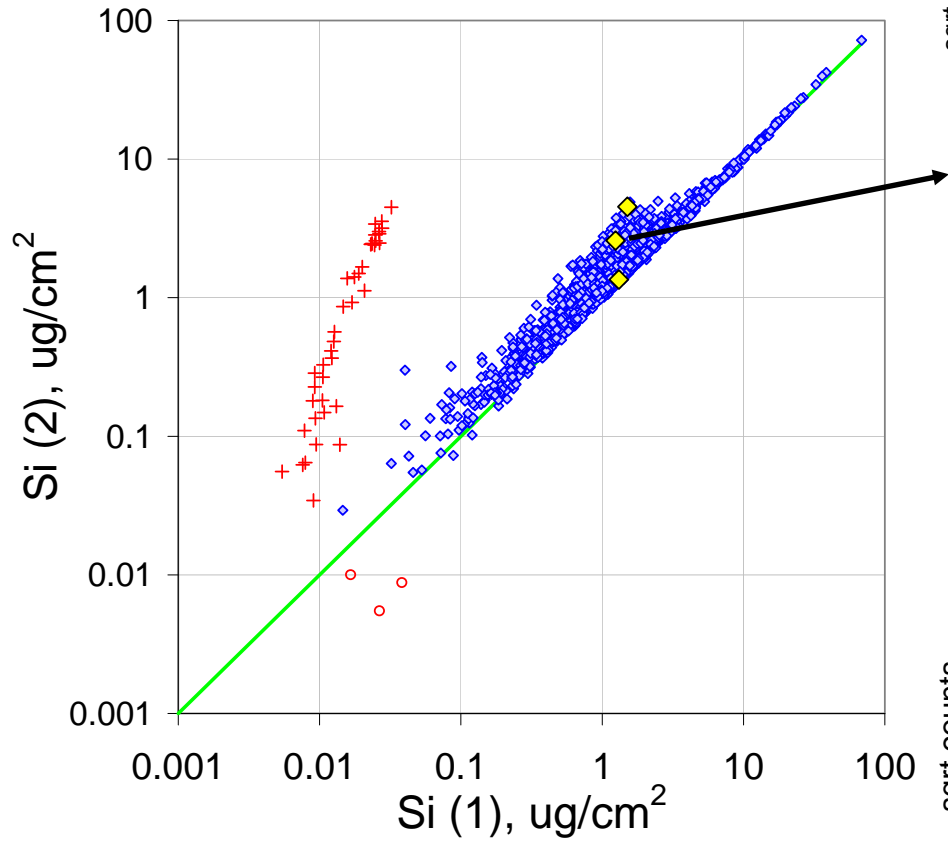
Teflon
blank

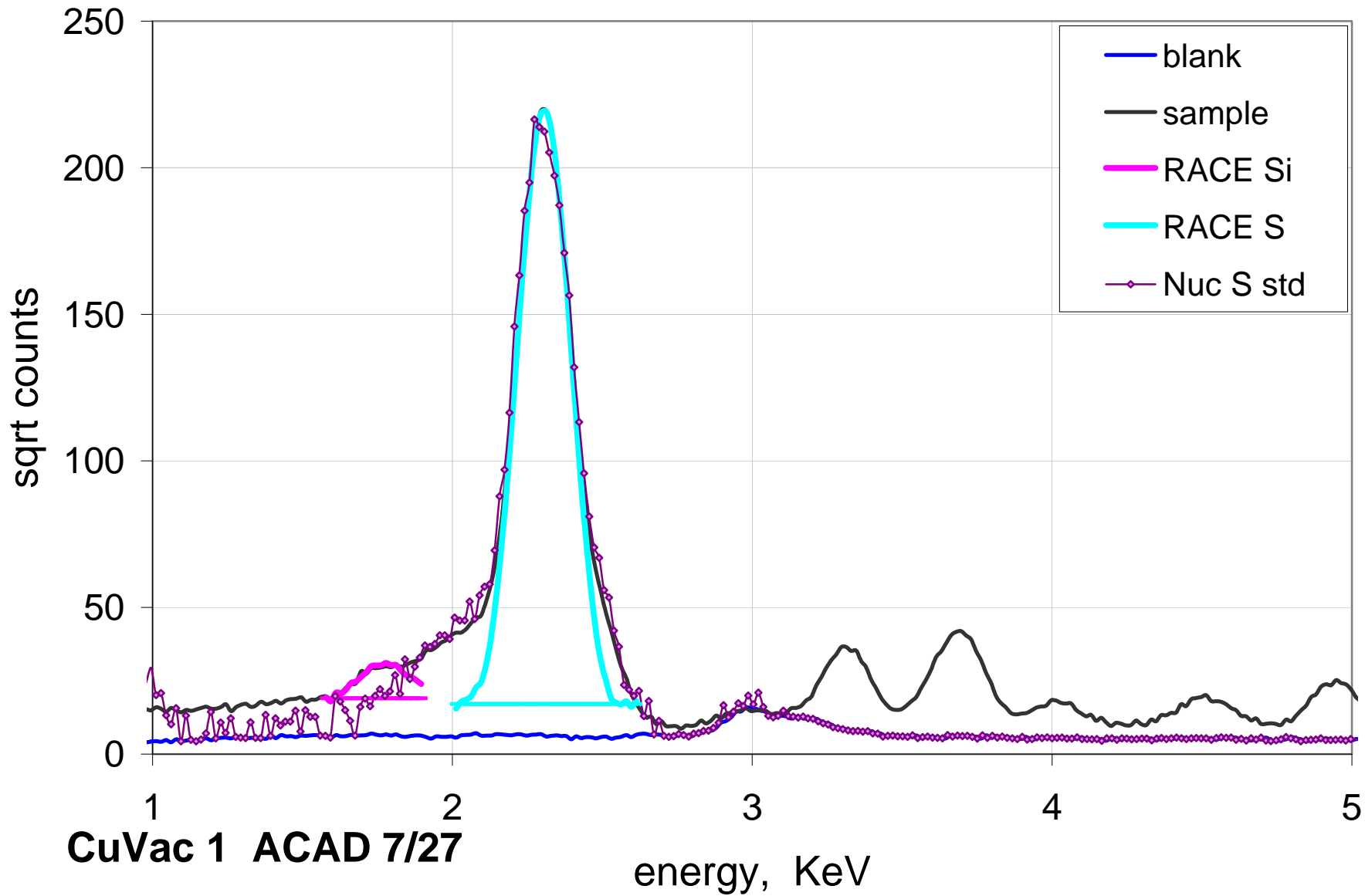


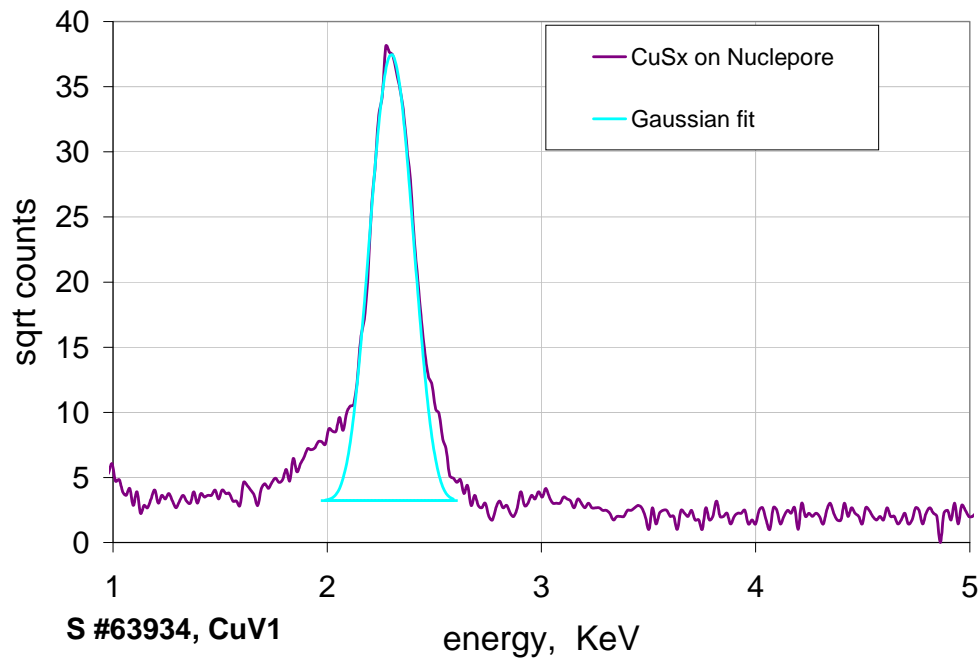
S standard
(on Mylar)



July 2005



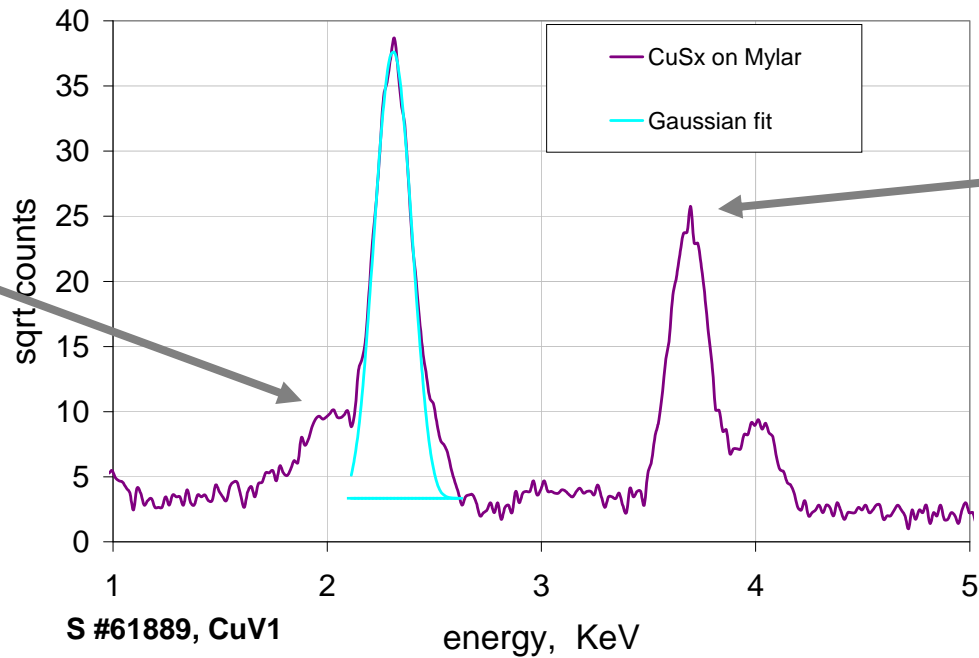




Nuclepore
backing, under
consideration

Si escape
peak:

$$\frac{3.69 \text{ KeV} - 1.74 \text{ KeV}}{1} = 1.95 \text{ KeV}$$

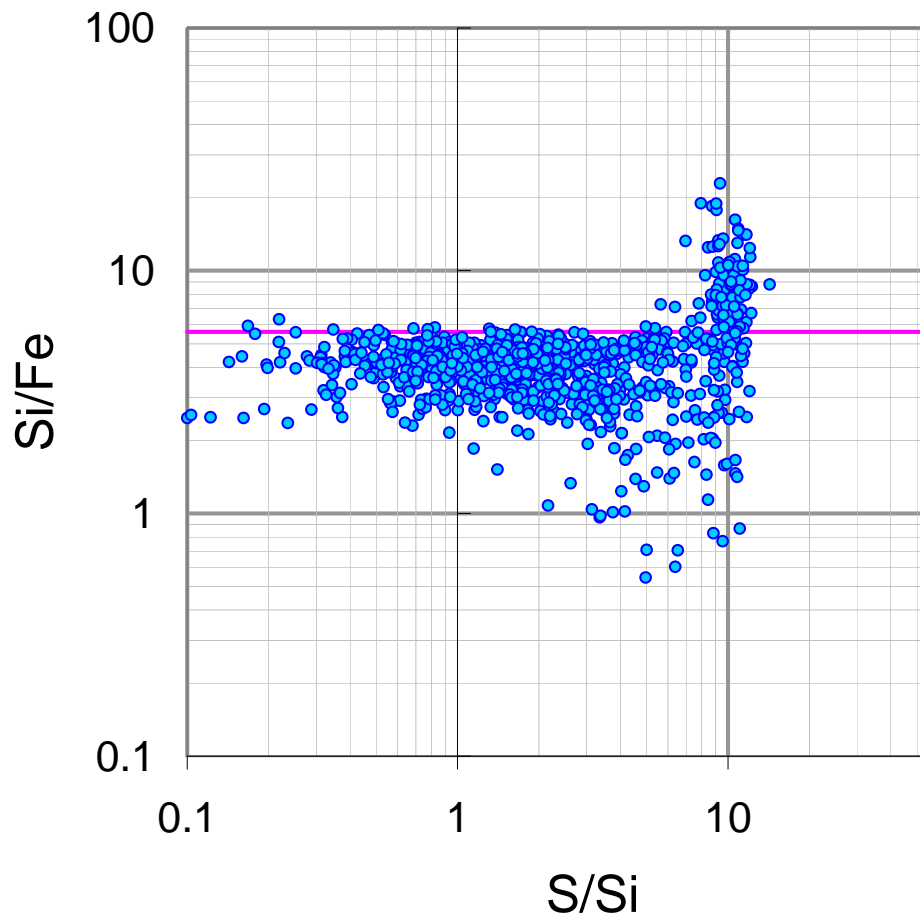


Ca

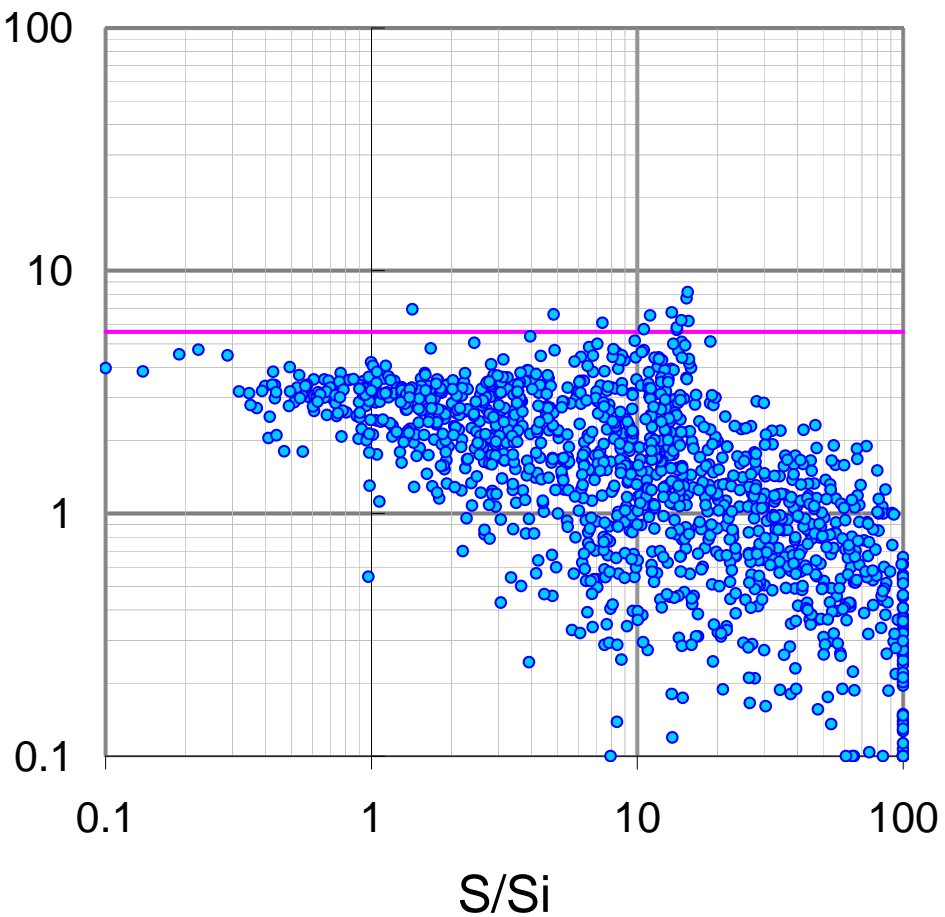
Mylar backing,
historical choice

July, 2004,

$\text{Fe} > 0.025 \text{ ug/m}^3$



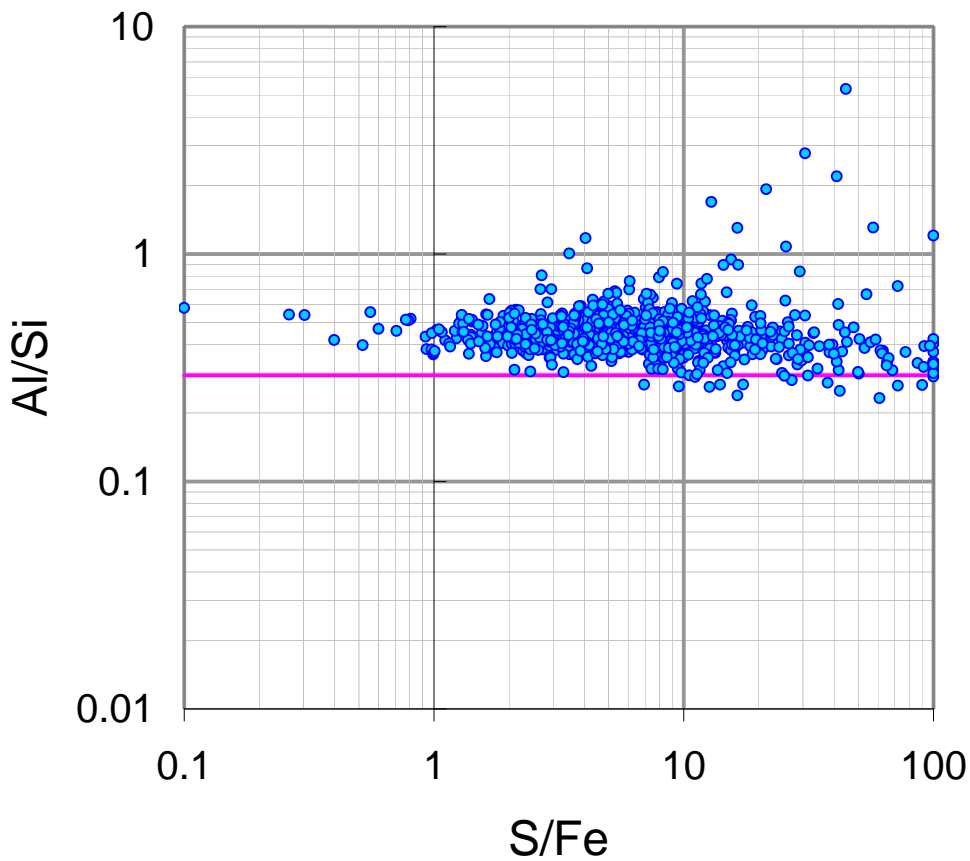
IMPROVE, Cu (He)



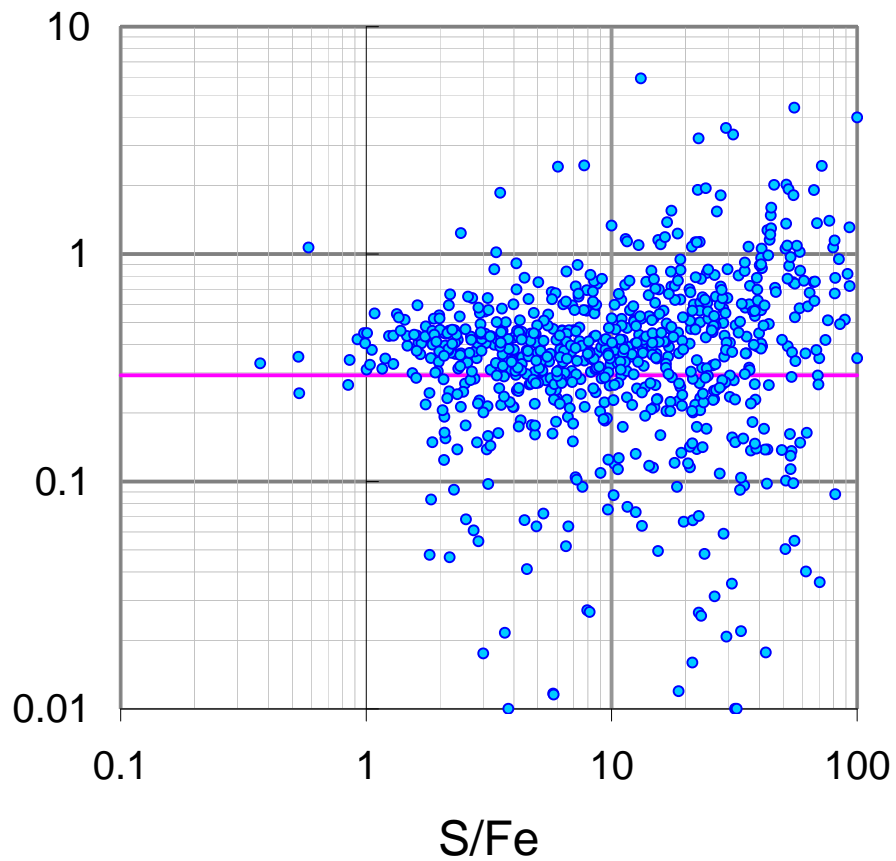
Chemical Speciation Network

July, 2004,

$\text{Fe} > 0.025 \text{ ug/m}^3$

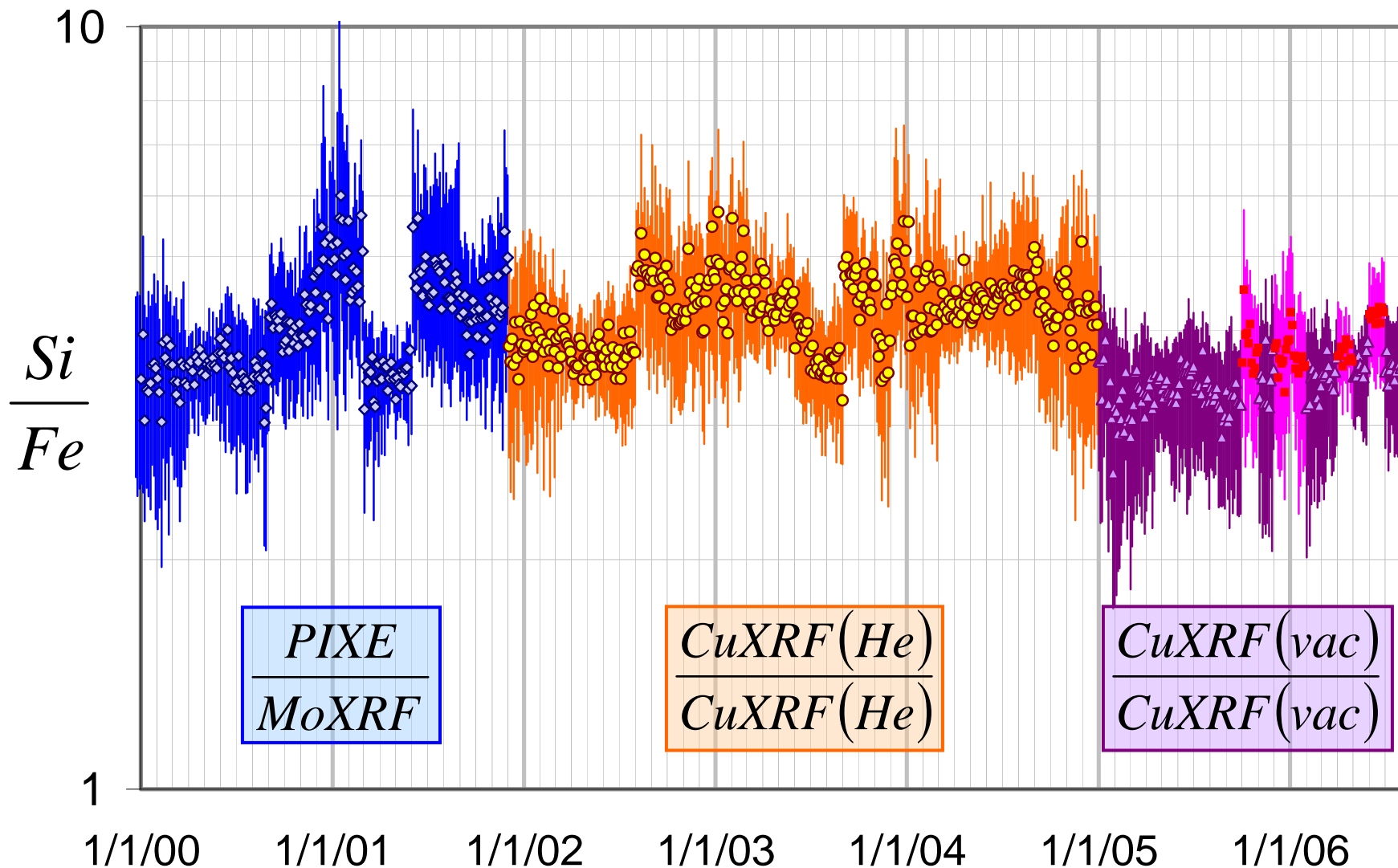


IMPROVE, Cu (He)

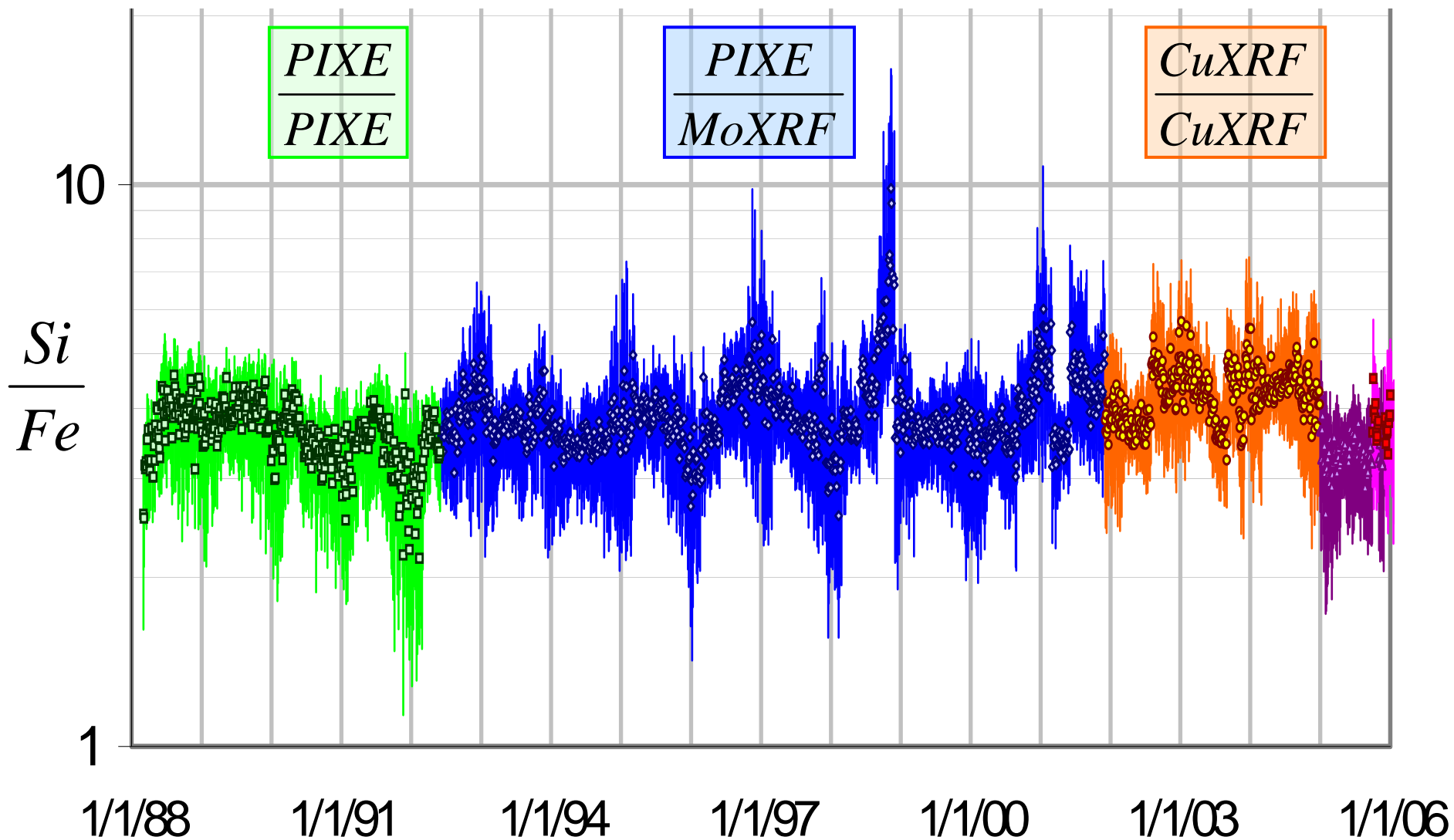


Chemical Speciation Network

MEDIAN AND INTERQUARTILE RANGE ENTIRE NETWORK, ALL DETECTED CONCENTRATIONS

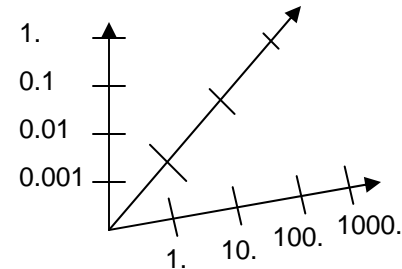


MEDIAN AND INTERQUARTILE RANGE
ENTIRE NETWORK, ALL DETECTED CONCENTRATIONS



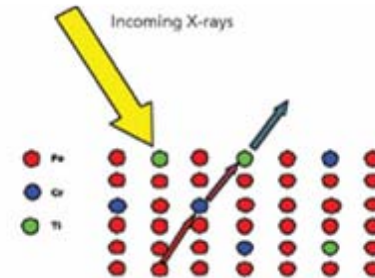
EDXRF is a bountiful measurement method.

It can simultaneously determine a dozen elements in each sample, at concentrations ranging over orders of magnitude. It can be automated to process samples of diverse compositions with a single set of calibration parameters.



Its bounty challenges calibration.

The disparate environments represented in the network span a huge volume in the space of concentrations. Calibration tolerances that are more than adequate in one region may yield unacceptable errors in another.



My personal conclusion: “Of whom much is given, much is expected.”

The user of routine EDXRF data must therefore accept some of the responsibility for assessing its quality.